QP Code: RM22MSC131	Reg.						AY 22
	No						



GIET UNIVERSITY, GUNUPUR - 765022

M. Sc. (Fourth Semester) Examinations, May - 2024

20PHPE402 - Condensed Matter & Materials Physics - 2 (Physics)

Maximum: 70 Marks Time: 3 hrs

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	(The figures in the right-hand margin indicate marks.)			
PART – A		$(2 \times 10 = 20 \text{ Marks})$		
Q.1	. Answer ALL Questions			
a.	What is meant by hysteresis in ferromagnetic magnetic materials?	CO1	K1	
1		th CO1	K1	
b.	What is anti-ferromagnetism? Draw the variation of susceptibility wi	tn cor	Kı	
	temperature for antiferromagnetic materials.			
c.	The magnetic susceptibility of silicon is $-4.0*10^{-5}$. What is the flux density an	nd ^{CO1}	K2	
	total magnetic moment per unit volume in a magnetic field of intensi	ty		
	10^5amp/m ?			
d.	Explain the terms spin waves and magnons	CO1	K1	
e.	What is dielectric polarization? Discuss different polarization mechanisms.	CO2	K1	
f.	What is a complex dielectric constant?	CO2	K 1	
g.	State Bragg's law? Why Xray is used for crystal diffraction.	CO3	K1	
h.	What is crystallization temperature	CO3	K1	
i.	Differentiate between SEM and TEM.	CO3	K1	
j.	What are the four 4 types of nanomaterials? Mention some applications of the	he ^{CO4}	K 1	
	nano- materials.			
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P	ART - B (1)	$0 \times 5 = 50 \text{ Marks}$		

PART – B $(10 \times 5 = 50 \text{ Marks})$

Answer ANY FIVE the questions			CO#	Blooms Level
2. a.	Describe briefly the domain theory of ferromagnetism. Derive Curie -Weiss law for their susceptibility.	10	CO1	K2
3.a.	Derive Curie's law for paramagnetic susceptibility quantum mechanically. How this theory removes Langevin's theory?	8	CO1	K2
b.	Discuss ferroelectric domains.	2	CO1	K1
4.	Discuss dipolar, ionic, and electronic polarizabilities. Plot a graph in total polarizability and frequency, clearly indicating the contributions of all the three polarizabilities.	10	CO2	K2
5.	Discuss Lorenz field in solid dielectric and hence derive the Clausius-Mossotti relation.	10	CO2	K2
6. a.	Block diagram of heat flux DSC.	3	CO3	K1
b.	If glass transition, melting temperature and crystallization temperature are put all together, how the graph is look like?	7	CO3	K2
7.a.	Discuss the structure of various forms of Carbon-based nano- materials and	7	CO4	K1
	their applications.			

b.	List out the properties of materials that are being affected by nano- meter size	3	CO4	K2
	of particles with suitable examples.			
8. a.	Discuss the density of states for 3D bulk material.	8	CO4	K2
b.	Draw the density of states vs energy and no of states vs energy graph in a	2	CO4	K2
	quantum wire			

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