



**GIET UNIVERSITY, GUNUPUR - 765022**  
**B. Tech (Fourth Semester Regular) Examinations, May - 2024**  
**22BCHE24001 - Material Science**  
**(Chemical)**

Time: 3 hrs

Maximum: 70 Marks

(The figures in the right hand margin indicate marks)

**PART - A****(2 x 5 = 10 Marks)**Q.1. Answer **ALL** questionsCO #      Blooms  
Level

- |  |     |    |
|--|-----|----|
| a. Differentiate unit cell and primitive cell. | CO1 | K2 |
| b. What is an isomorphous system?              | CO2 | K1 |
| c. What is Critical Cooling Rate (CCR)?        | CO3 | K1 |
| d. Write some objectives of Heat treatment?    | CO3 | K1 |
| e. Define cermet.                              | CO4 | K1 |

**PART - B****(15 x 4 = 60 Marks)**Answer ALL questionsMarks      CO #      Blooms  
Level

- |   |    |     |    |
|---|----|-----|----|
| 2. a. Aluminium has FCC structure and its density is 2700kg/m <sup>3</sup> . Calculate the unit cell dimension and atomic diameter. ( $A_w$ of Al = 26.98 g/mol).   | 8  | CO1 | K3 |
| b. Classify bonding in solids and explain briefly.  | 7  | CO1 | K2 |
| (OR)  |    |     |    |
| c. Describe and illustrate the edge and screw dislocations. Draw Burgers circuit to show magnitude and direction of Burgers vector on a crystal having edge dislocation.  | 10 | CO1 | K2 |
| d. Calculate the equilibrium number of vacancies per cubic meter for copper at 100° C. The energy for vacancy formation is 0.9 eV/atom; the atomic weight and density for copper are 63.5 g/mol and 8.45 g/cm <sup>3</sup> , respectively. Take Avogadro's number as 6.023x10 <sup>23</sup> atoms/mol. $k=8.62 \times 10^{-5}$ eV/atom. | 5  | CO1 | K3 |
| 3.a. Define non-equilibrium cooling. What is the impact of this cooling?  | 7  | CO2 | K1 |
| b. Explain the effect of cold working and hot working process on metals and write the advantages and disadvantages of both with example of both the processes.  | 8  | CO2 | K2 |
| (OR)  |    |     |    |
| c. Draw the Iron-Cementite (Fe-Fe <sub>3</sub> C) diagram & label the phase fields. Discuss in brief different reactions that take place in this system.  | 15 | CO2 | K2 |
| 4.a. What are the types of Heat treatment process, Explain briefly.   | 8  | CO3 | K2 |
| b. Define hardenability. Explain Jominy End-Quench Test.  | 7  | CO3 | K2 |
| (OR)  |    |     |    |
| c. What is meant by Normalizing? List the objectives of normalizing.  | 8  | CO3 | K2 |
| d. Explain recovery, recrystallization and grain growth with proper diagram.  | 7  | CO3 | K2 |
| 5.a. Explain the working principle of Ruby Laser with proper diagram.   | 8  | CO4 | K2 |

- |   |   |     |    |
|---|---|-----|----|
| b. What are Optical fibres? Explain the structure and working principle of optical fibre. | 7 | CO4 | K2 |
| (OR)  |   |     |    |
| c. Explain the working principle of He-Ne Laser with proper diagram.                      | 8 | CO4 | K2 |
| d. What are ceramic materials? Explain briefly different types of ceramics.               | 7 | CO4 | K2 |

--- End of Paper ---