QP Code: RM22MSC035	Reg.					
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## **GIET UNIVERSITY, GUNUPUR – 765022**

AR 22

M. Sc. (First Semester) Examinations, March - 2023

## 22CHPC104 - Molecular Spectroscopy-I

(Chemistry) Time: 3 hrs Maximum: 70 Marks (The figures in the right hand margin indicate marks.) PART - A  $(2 \times 10 = 20 \text{ Marks})$ CO Blooms Q.1. Answer all questions # Level CO2 K2 In terms of B where I is the moment of inertia, what is the energy,  $\Delta E$ , For the rigid rotor transition, J=4 J=5.7 $\rightarrow$ CO<sub>1</sub> b. Define Hooke's law. **K**1 c. Determine the ground state term symbol for Mn<sup>2+</sup>, V<sup>3+</sup>, and Fe<sup>2+</sup>. CO<sub>2</sub> K2 CO<sub>2</sub> d. Write note on Mutual exclusion principle. **K**1 CO<sub>1</sub> **K**1 e. Draw Morse Potential Curve. CO<sub>1</sub> f. What is molecular spectroscopy **K**1 Why N<sub>2</sub> molecule is inactive to rotational spectroscopy? CO<sub>3</sub> K2 g. CO3 Define Koopman's Theorem? **K**1 CO<sub>1</sub> **K**1 Define hyperfine coupling? CO<sub>4</sub> K2 What are the conditions for ESR active? PART - B  $(10 \times 5 = 50 \text{ Marks})$ Marks CO# Blooms Answer ANY FIVE questions Level CO<sub>2</sub> K2 2. a. Write note on Frank Condon Principles. 6 CO<sub>1</sub> K1 Describe the formula for finding wave length of emitted photon. Find the wave 4 length of photon when electron falls from n=2 to n=1 (For Hydrogen atom). CO<sub>2</sub> 3.a. Describe aim and features of sodium alkali spectra. K2 6 CO<sub>2</sub> K1 4 Write down the principle of Photo electron spectroscopy. CO<sub>1</sub> K1 6 4. a. Derive the expression of energy level in vibrational rotational transition. CO<sub>2</sub> K2 What is Raman spectroscopy and describe the structure illustration by Raman b. 4 spectroscopy. CO<sub>3</sub> K1 5.a. Calculate the number of vibrational mode for CO<sub>2</sub> and H<sub>2</sub>O molecule. 4 CO<sub>3</sub> K2 Derive the expression of rotational energy of rigid diatomic molecule? 6 CO<sub>1</sub> **K**1 What is Starck Effect? Explain in details about the splitting of rotational levels in 4 6. a. different category molecules? CO<sub>4</sub> K2 b. What is Fundamental Vibration? 6 CO<sub>4</sub> K1 8 Principle, instrumentation and Application of Auger electron spectroscopy CO<sub>3</sub> K1 What is Hot bands. b. CO<sub>2</sub> K2 8. a. Write Short notes on I. Zero-field splitting II. Kramer's Degeneracy CO<sub>1</sub> **K**1 Describe the basic principle of ESR spectroscopy? 4 --- End of Paper ---