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GIET UNIVERSITY, GUNUPUR – 765022

B. Tech (Fifth Semester - Regular) Examinations, December - 2022

BPCCH5010 - CHEMICAL PROCESS TECHNOLOGY

(Chemical Engineering)

			(Che	mical Engineering)					
Time: 3 hrs				Maximum: 70 Marks					
				Questions		_			
ъ.	DE A		t hand	l margin indicate marks.	10.34				
PART – A: (Multiple Choice Questions) Q.1. Answer ALL questions (1 x 10						l arks) PO#			
_		-	0		CO# CO1	PO1			
a.		act process of sulphuric acid manufacture yields acid of higher concentration		violds acids of lower concentration	001	101			
	i.	than chamber process.	ii.	yields acids of lower concentration than chamber process					
	iii.	is obsolete.	iv.	eliminates absorber					
b.		materials for 'Solvay Process' for manufa			CO1	PO2			
	i.	salt, limestone and coke or gas.	ii.	ammonia, salt and limestone.					
	iii.	ammonia limestone and coke.	iv.	none of these					
c.	Oil is			10.10 of 11.000	CO1	PO1			
•	i.	mixture of glycerides.	ii.	mixture of glycerides of fatty acids.					
	iii.	solid at normal temperature	iv.	ester of alcohols other than glycerine.					
		-		ester of acconors other than grycerine.					
d.	Unsa	turated oils compared to saturated oils ha	ave		CO1	PO2			
	i.	lower melting point & higher	ii.	higher melting point & higher					
		reactivity to oxygen.		reactivity to oxygen.					
	iii.	higher melting point & lower	iv.	higher melting point & lower					
	ъ.	reactivity to oxygen.		reactivity to oxygen.	CO2	PO1			
e.		dity of the fatty oil can be reduced by its			CO2	POI			
	i.	decoloration	ii.	hydrogenation					
	iii.	oxidation	iv.	purification	COL	DO1			
f.	-	s remove dirt by			CO2	PO1			
	i.	increasing the surface tension.	ii.	decreasing wettability.					
	iii.	supplying hydrophyllic group.	iv.	none of these					
g. Laboratory glass wares which reacts with hydrofluoric acid, are made of the						PO2			
	i.	Lead	ii.	Borosilicate					
	iii.	Soda lime	iv.	Alkali silicate					
h.	The ic	deal pulp for the manufacture of paper sh	ould h	nave high content.	CO4	PO1			
	i.	Cellulose	ii.	lignin					
	iii.	both (i) & (ii)	iv.	none of these					
i.				of	CO4	PO2			
	i.	alcohol	ii.	essential oil					
	iii.	fatty acids	iv.	massecuite					
j.	Soap	cannot be used with hard water, because			CO4	PO1			
J	i.	hard water contains sulphate.	ii.	they form insoluble calcium soaps which precipitate.					
	iii.	they attract back the removed dirt.	iv.	none of these.					
PART – B: (Short Answer Questions) (2 x 10 = 20 Marks)									
Q2. Answer ALL questions									
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a. Enumerate the three major industrial chemicals which comes under Chlor-Alkali Industry.

b.	Mention the chemical formula of washing soda.		CO1	PO2			
c.	Mention the properties and uses of chlorine.		CO2	PO2			
d.	Name the catalyst used for DCDA process for sulphuric acid manufacture and mention its characteristics?						
e.	Write down the reactions occurred at cathode and anode for a mercury cell used for the production of NaOH.						
f.	Identify the different sources of raw material that can used in Paper Industry.		CO2	PO1			
g.	What is the roll of Antioxidant in the Oil industry?		CO3				
h.	Write the chemical formula of ABS? What is its application?		CO3				
i.	Mention some of the principal compound which is used in white Pigment.						
j.	What are the by-products of sugar industry and its uses?		CO4	PO3			
PA	PART – C: (Long Answer Questions) (10 x 4 =						
Ans	wer ALL questions	Marks	CO#	PO#			
3.a.	Discuss in detail, the manufacture of soda ash by Solvay process giving emphasis on chemical reactions?	8	CO1	PO1			
b.	Analyze some of the Process Equipment Symbols which are used for Evaporation, Fluid-solid Contacting.	2	CO1	PO2			
	(OR)						
c.	Illustrate the synthetic Hydrochloric acid production with a neat flow sheet?	8	CO1	PO3			
d.	Analyze some of the Process Equipment Symbols which are used for Extraction and Fluid Handling.	2	CO1	PO2			
4.a.	Explain in details about continuous process for fatty acids, soaps and glycerine with a neat flow chart?	8	CO2	PO1			
b.	Analyse about Fat Splitting and Saponification by mentioning the chemical reaction.	2	CO2	PO1			
(OR)							
c.	With the help of neat flow sheet describe the sulphate process for the preparation of wood pulp	8	CO2	PO1			
d.	How and which chemical is used for Bleaching of Pulp.	2	CO2	PO2			
5.a.	Describe about Indian Sugar Industry mentioning about Extraction of Sugar form sugar cane with neat flow sheet	8	CO3	PO2			
b.	How massecuite is converted to blackstrap molasses?	2	CO3	PO3			
	(OR)						
c.	What are the raw materials for preparation of glass?	2	CO3	PO1			
d.	What are the manufacturing steps involve in preparation of glass?	8	CO3	PO2			
6.a.	Discuss in details with a neat flow sheet the production of polyethylene by low pressure Ziegler's process. Mention its properties and uses.	6	CO4	PO1			
b.	Explain about the chemistry of Polycondensation Process.	4	CO4	PO3			
	(OR)						
c.	Examine the chemistry, production and uses of cellulose .Mention the production process of 6,6-nylon	6	CO4	PO1			
d.	Classify the Polymer On the basis of physical structure	4	CO4	PO2			
	End of Paper						