QP Code: RM20MSC093	Reg.						AR 20
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GIET UNIVERSITY, GUNUPUR – 765022

M. Sc (First Semester) Examinations, May – 2021

20CHPC101 - ORGANIC CHEMISTRY - I

(Chemistry)

Time: 2 hrs Maximum: 50 Marks

 $PART - A (2 \times 10 = 20 \text{ Marks})$

Q.1. Answer ALL the questions

- a. Tropylium bromide gives a yellow precipitate with AgNO₃ solution, while bromobenzene does not. Reason out.
- b. State Huckel's rule with two examples.
- c. Arrange the following compounds according to increasing pK_a values and give reason: Dichloro acetic acid, acetic acid, propionic acid, trichloro acetic acid
- d. Classify as hard or soft acids or bases: H₂O, CN⁻, Pt⁺², BH₃
- e. Write Hammett's equation and explain the terms involved.
- f. Distinguish configuration from conformation.
- g. Assign E/Z notation:

i)
$$H_5C_2$$
 $COOMe$ H_5C_2 C_6H_5 CH_5 CH_5 CH_5

- h. How does a racemic mixture differ from a meso compound?
- i. What are ambident nucleophiles? Give two examples.
- j. Explain Walden inversion.

PART - B (6 x 5 = 30 Marks)

Answer ANY FIVE questions			
2.	Explain any three methods to generate carbenes.	6	
3.	Discuss kinetically and thermodynamically controlled product with an example.	4+2	
4.	Discuss the stereochemistry of cis and trans decalin	6	
5.	Assign R/S configuration:	6	

- 6. Explain the following observations.
 - i) Allyl chloride undergoes substitution reaction through S_N1 mechanism while n-propyl chloride undergoes through S_N2 pathway.

6

6

- ii) Vinyl chloride and chloro benzene do not favour nucleophilic substitution reaction.
- 7. Discuss the effect of structure of the substrate and the nucleophile on S_N1 and S_N2 6 mechanisms.
- 8. Write a brief note on Pearson's HSAB theory.

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