Registration No. :		The state of the s							
Total number of printed pages – 2							B.Tech		
							F	PCCH 420	2

Special Examination – 2012 CHEMICAL PROCESS TECHNOLOGY

Full Marks - 70

Time - 3 Hours

Answer Question No. 1 which is compulsory and any **five** from the rest.

The figures in the right-hand margin indicate marks.

1. Answer the following questions:

2×10

- (a) Mention the properties and uses of caustic soda.
- (b) Mention the properties and uses of hydrochloric acid.
- (c) Why the hydrogenation of oil is necessary?
- (d) Write the chemical reactions involved for the soap production.
- (e) Name the chemicals recovered from black liquor.
- (f) Why are pigments used? Give examples of coloured and white pigments.
- (g) Write the formula and properties of sucrose.
- (h) What is dialdehyde starch? Mention the chemical reactions.
- (i) What is fiber? Give some examples.
- (j) Write the structure and properties of polyethylene.
- Discuss in detail the Solvay process for soda ash production with a neat flow sheet. Also discuss the major engineering problems associated with its production.

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- Discuss in detail the Contact process for sulfuric acid production with a neat flow sheet. Also discuss the major engineering problems associated with its production.
- 4. (a) Write the chemical composition and physical properties of vegetable oils.

4

(b) Discuss the methods of extracting vegetable oils.

6

- 5. Discuss in detail with a neat flow sheet the Kraft process of pulp production giving emphasis on major engineering problems. 3+3+4
- 6. Discuss in detail with a neat flow sheet the production of ethyl alcohol by fermentation process giving emphasis on major engineering problems. 3+3+4
- 7. (a) Write the properties and uses of phenol formaldehyde.
 - (b) Discuss the production of phenol formaldehyde with a neat flow sheet. Also mention the chemical reactions.
- 8. Write short notes on any two:

5×2

- (a) Synthetic detergents
- (b) Dextrin
- (c) Natural dyes
- (d) Polyvinyl acetate.

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