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Total number of printed pages – 2

B. Tech
PCCH 4302

Fifth Semester (Back / Special) Examination – 2013

MASS TRANSFER – I

BRANCH : CHEM

QUESTION CODE : D 257

Full Marks – 70

Time : 3 Hours

*Answer Question No. 1 which is compulsory and any **five** from the rest.*

The figures in the right-hand margin indicate marks.

Assume suitable notations and any missing data wherever necessary.

answer all parts of a question at a place.

1. Answer the following questions : 2 × 10
- Write and explain Fick's first law of diffusion.
 - On what factors does the mass transfer rate between two fluid phases depend ?
 - Write the Gilliland's equation.
 - For what value of relative volatility, the separation becomes uneconomical ?
 - Discuss the importance of minimum reflux ratio.
 - When steam distillation is used ?
 - Mention the characteristics of a good absorbent.
 - Write the expressions used for the estimation of packed height based on gas film and liquid film.
 - Define dry bulb, wet bulb, dew point, and adiabatic saturation temperatures.
 - Write the Lewis relation for air-water system.
2. In an O_2 - N_2 mixture at 10 atm and $25^\circ C$, the concentrations of O_2 at two places of 0.18 cm apart are 11 and 22 vol % respectively. Calculate the rate of diffusion of O_2 in $gm/cm^2 \cdot hr$ for the case of unicomponent diffusion (N_2 is non-diffusing). Value of diffusivity is $0.18 cm^2/s$. 10



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3. Discuss the concept of VLE for binary systems with a neat PTxy diagram. Also discuss the effect of T on Pxy diagram. 10
4. (a) Write the assumptions made in McCabe Thiele's method. 4
 (b) Derive the equation of q-line. 6
5. A liquid mixture containing 50 mol % heptane (A) and 50 mol % octane (B), is to be continuously flash vaporized at 1 atm pressure to vaporize 60 mol % of the feed. What will be the composition of the vapour and liquid in the separator for an equilibrium stage ? 10

Data :

T (°C)	98.5	105	110	115	120	125.5
Vapour pressure of A (mm Hg)	760	940	1050	1200	1350	1540
Vapour pressure of B (mm Hg)	333	417	484	561	650	760

6. 5500 kg/hr of a SO₂-air mixture containing 5 % by volume SO₂ is to be scrubbed with 2 × 10⁵ kg/hr of water in a packed tower. The exit concentration of SO₂ is reduced to 0.20 %. The tower operates at 1 atm. The equilibrium relationship is given by: Y = 30 X where, Y = mole SO₂/mole air, and X = mole SO₂/mole water. If the packed height of the tower is 450 cm, estimate the height of transfer unit (HTU). 10
7. (a) Discuss in detail the construction and working of a spray tower. 5
 (b) Discuss in detail the construction and working of a sling psychrometer. 5
8. Write short notes on any **two** of the following : 5×2
 - (a) Stefan's method for determination of diffusivity
 - (b) Minimum irrigation rate
 - (c) Optimum reflux ratio
 - (d) Spray pond.

