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Total number of printed pages – 2

B. Tech
PEBT 5304

Fifth Semester (Back/Special) Examination – 2013
BIOCHEMICAL REACTION ENGINEERING

BRANCH : BIOTECH
QUESTION CODE : D321

Full Marks – 70

Time : 3 Hours

Answer Question No. 1 which is compulsory and any **five** from the rest.

The figures in the right-hand margin indicate marks.

1. Answer the following questions : 2×10
- Define space time and space velocity.
 - Write the difference between order and molecularity of reactions.
 - Write the difference between homogeneous and heterogeneous reactions.
 - What are the advantages of fluidized bed reactors over packed bed reactors ?
 - What do you mean by psychometric chart ? Write the use of psychometric chart.
 - Write the conditions where Monod's growth model is not applicable.
 - What is critical dilution rate ? What happens to specific growth rate at critical dilution ?
 - On doubling the concentration of reactant, the rate of reaction triples. Find the reaction order.
 - Consider an isothermal gas-phase reaction $A \rightarrow 2R$. Calculate Δ_A of the reaction.
 - What do you mean by performance equation ?
2. (a) Derive an expression for velocity of an enzyme catalyzed reaction under "competitive inhibition". 5
- (b) Write short note on Monod's model of growth kinetics. 5
3. (a) Derive the performance equation of a PFR operating condition in ideal. 5
- (b) Derive the performance equation for reactor containing porous catalyst particle. 5
4. Discuss, in detail, the integral method of data analysis for irreversible unimolecular-type first order and biomolecular second order reaction for a varying volume batch reactor. 10

P.T.O.

5. (a) From a series of batch runs with a constant enzyme concentration, the following initial rate data were obtained as a function of initial substrate concentration. Evaluate the Michaelis-Menten kinetic parameters by employing the Langmuir plot, the Lineweaver-Burk plot, and the Eadie-Hofstee plot. 10

The following data were generated :

S_0 (nM/L)	V_0 (mM/L min)
1	0.2
2	0.22
3	0.3
5	0.45
7	0.41
10	0.5
15	0.4
20	0.33

6. (a) Write short note on PFR in series and parallel arrangement.
 (b) An aqueous feed of A and B (400 liter/min, 100 mmol A/liter, 200 mmol B/liter) is to be converted to product in a plug flow reactor. The kinetics of the reaction is represented by
- $$A + B \rightarrow R, \quad -r_A = 200C_A C_B \text{ mol/liter} \cdot \text{min}$$
- Find the volume of reactor needed for 99.9% conversion of A to product. 5
7. (a) Derive the equation for analysis of data by integral method for a zero order reaction operating in constant volume batch reactor. 5
 (b) Milk is pasteurized if it is heated to 63°C for 30 min, but if it is heated to 74°C it only needs 15s for the same result. Find the activation energy of this sterilization process. 5
8. Answer any **two** of the following : 5×2
- (a) Write a short note on autocatalytic reaction.
 (b) Derive the equation for Michaelis-Menten kinetics.
 (c) Derive the expression for Recycle reactor.
 (d) Write short note on Briggs-Haldane relationship.