Registration No.:							
Total number of pr	inted	page	es – 2				B. Tech.
							PCCH 4202

Third Semester Examination – 2012-13 CHEMICAL PROCESS TECHNOLOGY

Full Marks - 70

Time: 3 Hours

Answer Question No. 1 which is compulsory and any five from the rest.

The figures in the right-hand margin indicate marks.

1. Answer the following questions:

2×10

- (a) Mention the properties and uses of chlorine.
- (b) Name the catalyst used for DCDA process for sulphuric acid manufacture and mention its characteristics.
- (c) With a rough diagram, write the mechanism of cleaning of grease from a surface by the use of soaps and detergents.
- (d) Write the raw materials for pulp production and write the types of paper products.
- (e) Mention the temperature range under which mercury electrolytic cells are operated.
- (f) What is dextrin? Write its chemical formula.
- (g) What are mash and slops?
- (h) What is the use of antioxidants in oil industries?
- (i) Mention the quantitative requirements for one ton of 100% alcohol.
- (j) Write the properties and uses of polyvinyl acetate.

2.	Dis	cuss in detail the electrolytic process for chlorine-caustic soda p	roduction
	with	n a neat flow sheet. Also discuss the major engineering problems a	ssociated
	with	n their production.	4+3+3
3.	(a)	Discuss briefly about hydrogenation of oil.	5
	(b)	Discuss the methods for detergent manufacture.	5
4.	dige	cuss in detail the Kraft pulp process with a neat flow sheet giving emestion, bleaching, and recovery of chemicals. Also discuss the gineering problems.	•
5.		cuss in detail with a neat flow sheet the extraction of sucrose from sucrystalline sugar production giving emphasis on major engineering pr	•
6.		cuss in detail with a neat flow sheet the production of polyethyler ssure Ziegler process. Mention its properties and uses.	ne by low 4+4+2
7.	(a)	Write the properties and uses of PVC.	2
	(b)	Discuss the production of PVC with a neat flow sheet. Also me engineering problems.	ntion the 4+4
8.	Wri	te short notes on any two:	5×2
	(a)	Chemical reactions for Solvay process	
	(b)	Pigments and dyes	
	(c)	Industrial alcohol	
	(d)	Phenol formaldehyde.	