Registration No:											
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Total Number of Pages: 02

B.TECH PBT3I101

3rd Semester Regular Examination 2016-17 BIOCHEMISTRY

BRANCH: BIOTECH Time: 3 Hours Max Marks: 100

Q.CODE: Y526

Answer Part-A which is compulsory and any four from Part-B.

The figures in the right hand margin indicate marks.

Part – A (Answer all the questions)

Q1 Answer the following questions: multiple type or dash fill up type

(2 x 10)

- Keratan sulphate is found in abundance in (A) Heart muscle (B) Liver
 (C) Adrenal cortex (D) Cornea
- **b)** Invert sugar is (A) Lactose (B) Sucrose (C) Hydrolytic products of sucrose (D) Fructose
- c) Specific gravity of urine increases in (A) Diabetes mellitus (B) Chronic glomerulonephritis (C) Compulsive polydypsia (D) Hypercalcemia
- d) The number of isomers of glucose is (A) 4 (B) 8 (C) 12 (D) 16
- e) Pyruvate₂₁₀dehydrogenase₁₀ complex and α-ketoglutarate₁₀ dehydrogenase complex require the following for their oxidative decarboxylation: (A) COASH and Lipoic acid (B) NAD+ and FAD (C) COASH and TPP (D) COASH, TPP,NAD+,FAD, Lipoate
- f) The following is an enzyme required for glycolysis: (A) Pyruvate kinase (B) Pyruvate carboxylase (C) Glucose-6-phosphatose (D) Glycerokinase
- Pentose production is increased in (A) HMP shunt (B) Uromic acid pathway
 (C) EM pathway (D) TCA cycle
- following condition: (A) Diabetes insipidus (B) Diabetes Mellitus (C) Hypothyroidism (D) Liver diseases
- i) In Lineweaver-Burk plot, the y-intercept represents (A) Vmax (B) Km (C) Km
 (D) 1/Km
- j) In competitive inhibition, the inhibitor (A) Competes with the enzyme (B) Irreversibly binds with the enzyme (C) Binds with the substrate (D) Competes with the substrate

Q2 Answer the following questions: Short answer type

 (2×10)

- **a)** Draw the structure of glucose.
- **b)** Saturated lipids are solid where as unsaturated are liquids-Explain.
- c) Differentiate between reaction intermediate and transition state.
- d) What do you mean by T_m of DNA?
- e) Differentiate between de novo and salvage pathway of nucleotide synthesis.
- f) What do you mean by energy coupling in biological reactions?

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210	g) h) i) j)	What is prosthetic group Name the chemical name Define K _m What is CA cycle?		· 210	210	210		21
Q3	a) b)	Part – Explain the process of pro	photpphosphor				(10) (5)	
Q4 210	a) b)	Explain different types of Explain Ramachandran	•	oition with their k	zinetics.	210	(10) (5)	21
Q5	a) b)	Explain photo phosphyr Explain precursors of va	•		hesis.		(10) (5)	
Q6	a) b)	Explain dark reaction w Explain the titration cur	-	210	210	210	(10) (5)	21
Q7	a) b)	Explain levels of protein Explain mechanism of e					(10) (5)	
Q8	a)	Explain glycolysis. Ho oxidation of one glucos		will be forme	d from comple	ete	(10)	
	b)	Explain the structure of					(5)	
210 Q9		_	RNA. of lipids. How it		alpha oxidation	²¹⁰	(5) (10) (5)	21
	a)	Explain the structure of Explain Beta oxidation of	RNA. of lipids. How it	is different from			(10)	21
	a) b)	Explain the structure of Explain Beta oxidation of	RNA. of lipids. How it	is different from			(10)	21
Q9	a) b)	Explain the structure of Explain Beta oxidation of Explain mechanism of e	RNA. of lipids. How it	is different from			(10)	21
Q9	a) b)	Explain the structure of Explain Beta oxidation of Explain mechanism of e	RNA. 210 of lipids. How it energy transfer	is different from	alpha oxidation		(10)	21
Q9	a) b)	Explain the structure of Explain Beta oxidation of Explain mechanism of e	RNA. 210 of lipids. How it energy transfer	is different from from ATP.	alpha oxidation	1 . 210	(10)	21

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