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No. of Pages : 2

B.Tech
BE2103

First Semester Examinations, 2012-13

THERMODYNAMICS

Full Marks: 70

Time: 3 hrs

Steam table and Molliers chart are allowed

Answer any SIX including Q1 which is compulsory

Figures in the right hand margin indicate marks

- Q1. Answer the followings 2x10
- a) Define thermodynamic system and classify them.
 - b) What do you mean by thermodynamic equilibrium and how it differs from thermal equilibrium.
 - c) Why it is incorrect to say 'system contains heat'.
 - d) Define moving boundary work, gravitational work, and acceleration work.
 - e) What is internal latent heat and how it is related to enthalpy of vaporization.
 - f) Prove that for a constant pressure process $dq=dh$
 - g) How heat is transferred during conduction?
 - h) Define a throttling process and write the steady state equation for the same.
 - i) What are mass fraction and mole fraction and how they are related?
 - j) What is a heat pump and how does it differ from a refrigerator?
- Q2. a) Prove that all reversible engines operating between same temperature limits are equally efficient. 5
- b) In a throttling calorimeter the steam is admitted at 10 bar, it is throttled to atmospheric pressure and 110°C. Determine the dryness fraction of steam. 5
- Q3. a) Prove that violation of Kelvin Planck statement leads to violation of Clausius statement. 4
- b) 0.4 kg of air at 6.0 bar receives an amount of heat at constant volume so that its temperature rises from 383 K to 923 K. It is then expanded polytropically according to $Pv^{1.32} = \text{Constant}$ to initial temperature and finally it is compressed isothermally to its

original volume. Calculate a) pressure at end state, b) work transfer and heat transfer during each process. 6

Q4. A compressor takes air at 100 kN/m^2 and delivers the same at 550 kN/m^2 . The compressor discharges 16 m^3 of air/min. The density of air at inlet and exit are 1.25 kg/m^3 and 5 kg/m^3 . The power of the motor driving the compressor is 40 kW . The heat lost to the cooling water circulated around the compressor is 30 kJ/kg of air passing through the compressor. Neglecting the PE and KE, determine the change in specific internal energy. 10

Q5. Draw a schematic of a thermal power plant, describe the different components including the accessories of it and discuss how the electric power is generated.? 10

Q6. A vessel of 6.0 m^3 capacity contains two gases A and B in proportion of 45% and 55% respectively at 30°C . If the gas constant R for the gases is 0.288 kJ/kg-K and 0.295 kJ/kg-K and total weight of mixture is 2.0 kg , calculate a) the partial pressure, b) the total pressure, c) the mean value of R for the mixture. 10

Q7. A reversible heat engine works between three reservoirs A,B,C. The engine absorbs an equal amount of heat from the thermal reservoir A and B kept at temperatures T_A and T_B respectively and rejects heat to the thermal reservoir C at T_C . The efficiency of the engine is 'n' times that of the reversible engine operating between the thermal reservoir A and C. Prove that

$$T_A/T_B = (2n-1) + 2(1-n)T_A/T_C \quad 10$$

Q8. Write short notes on the following. 2.5X4

- a) Entropy generation?
- b) Gas thermometer
- c) Combined mode of heat transfer
- d) Critical point and triple point of water