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Total number of printed pages – 3

B. Tech
BE 2103

First Semester Regular Examination – 2014

THERMODYNAMICS

BRANCH : B. TECH

QUESTION CODE : H 457

Full Marks – 70

Time : 3 Hours



Answer Question No. 1 which is compulsory and any **five** from the rest.

The figures in the right-hand margin indicate marks

1. Answer the following questions : 2 × 10
- (a) What do you mean by Quasi-static process ?
 - (b) What is the difference between universal gas constant and characteristic gas constant ?
 - (c) What is thermocouple ? On what principle, thermocouple works ?
 - (d) Show that work is a path function, not a property.
 - (e) Find the enthalpy, entropy at 1.4 Mpa, 400°C.
 - (f) Draw the p-v and T-s diagram for a reversed carnot cycle.
 - (g) Convert 40 cm Hg vacuum pressure into pressure in kPa format.
 - (h) What is COP ? Relate COP of refrigerator with that of heat pump.
 - (i) Name four mountings used in steam power plant.
 - (j) Draw constant pressure process and constant volume process on same T-s diagram.
2. (a) Show that energy is a property of a system. 4
- (b) A fluid at 0.7 bar occupying 0.09m³ is compressed reversibly to a pressure of 3.5 bar according to a $p v^n = \text{constant}$. The fluid is then heated reversibly

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at constant volume until the pressure is 4 bar, the specific volume is then $0.5 \text{ m}^3/\text{kg}$. A reversible expansion according to a law $pV^2 = \text{constant}$, restores the fluid to its initial state. Sketch the cycle on p-V diagram and calculate

- (i) the mass of the fluid present
- (ii) the value of "n" in the first process
- (iii) the net work of the cycle.

6

3. (a) What do you mean by an Adiabatic process? Derive the p-V relation for adiabatic process, i.e. $pv^k = \text{constant}$ where, adiabatic index, $k = c_p/c_v$

5

(b) A gas of mass 1.5 kg undergoes a quasi-static expansion which follows a relationship $p = a + bV$ where a and b are constants. The initial and final pressures are 1000 kPa and 200 kPa respectively and the corresponding volumes are 0.20 m^3 and 1.20 m^3 . The specific internal energy of the gas is given by the relation $u = 1.5pv - 85 \text{ kJ/kg}$ where p is kPa and v is in m^3/kg . Calculate the net heat transfer and the maximum internal energy of the gas attained during expansion.

5

4. (a) Write down the mass conservation equation and steady flow energy equation single stream entering and single stream leaving open system having

4

(b) A turbine operating under steady flow conditions receives steam at the following state: Pressure = 13 bar, specific volume = $0.13 \text{ m}^3/\text{kg}$, Specific internal energy = 2300 kJ/kg, Velocity = 30 m/s. The state of the steam leaving the turbine is as follows: Pressure = 0.3 bar, Specific volume = $4.3 \text{ m}^3/\text{kg}$, specific internal energy = 1500 kJ/kg, Velocity = 90 m/s. Heat rejected to the surrounding at the rate of 0.25 kW and the rate of steam flow through turbine is 0.4 kg/s. Calculate the power developed by the turbine.

6

5. (a) Establish the equivalence of Kelvin-Planck and Clausius statements.

4

(b) A heat pump is to be used to heat a house in winter and then reversed to cool the house in summer. The interior temperature is to be maintained at 20°C . Heat transfer through the walls and roof is estimated to be 500 J/s per degree temperature difference between the inside and outside.

- (i) if the outside temperature in winter is 5°C what is the minimum power required to drive the heat pump?
- (ii) What is the maximum outer temperature for which the inside can be maintained at 20°C . 6
6. (a) What do you mean by entropy principle? State Clausius theorem. 4
- (b) 0.5 kg of air is compressed at constant volume from 0.5MPa, 0.2m^3 , 100°C to 0.05m^3 , 160°C . Determine the change in internal energy, change in enthalpy and change in entropy. 6
7. (a) Draw h-s diagram for water and indicate the following on the same 4
- (i) saturated liquid line
 - (ii) saturated vapor line
 - (iii) critical point
 - (iv) constant pressure line
 - (v) constant temperature line
 - (vi) constant quality line
- (b) A rigid tank of volume 3m^3 contains 5 kg of wet steam at a pressure of 200 kPa. The tank is heated until the steam becomes dry saturated. Determine the final pressure, the heat transfer to the tank and entropy change. 6
8. Write short notes on any two : 5×2
- (a) Spark ignition engine and Compression ignition engine
 - (b) Reciprocating type air compressor and centrifugal type air compressor
 - (c) Refrigerator and heat pump.

