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ech **HSSM 3402** 

## Eighth Semester Regular / Back Examination - 2015 **ENVIRONMENTAL ENGINEERING**

BRANCH(S): CSE, FASHION, IT, MECH, MM, MME, TEXTILE **QUESTION CODE: J 107** 

Full Marks - 70

Time: 3 Hours

Answer Question No. 1 which is compulsory and any five from the rest. The figures in the right-hand margin indicate marks.

Answer the following questions: 1.

 $2 \times 10$ 

- Name the three biotic components of an ecosystem. (a)
- Write the environmental importance of nitrogen cycle. (b)
- What are different types of alkalinities? The pH of a given water sample (c) is 7.6. Find out the noncarbonated alkalinity in mg/L as CaCO3.
- Mention the drinking water quality standard (IS10500) both desirable and (d) permissible for colour and fluoride.
- Differentiate between batch reactor and flow reactor. (e)
- What are the four types of equalization in waste water pre treatment? (f)
- Calculate the Combustion Efficiency of an incinerator where the outlet gas (g) concentration for CO is 0.1% and inlet CO2 is 25.8%.
- Enlist the important properties of hazardous wastes. (h)
- Explain the sound pressure level and its unit. (i)
- Differentiate between Lifecycle Assessment (LCA) and Environmental (i) Impact Assessment (EIA).

- Give a classification of types of ecosystems with suitable examples. Explain the energy flow in an ecosystem as an ecosystem process with the help of an energy flow model.
- Explain the concept of source-path-receiver with respect to noise pollution. If the
  average day time noise power level in an industrial area is 67-dB and night time
  noise power level is 55 dB, find out the equivalent noise power level in that area.
   Assume day time as 6 AM- 9 PM and night time as 9 PM-6 AM.
- (a) Discuss the operation of nitrogen cycle in nature involving the important steps in it.
  - (b) Mention the various environmental laws which have been enacted in our country to tackle environmental policition. Name the statutory bodies dealing with various environmental laws in our country.
- (a) Explain the chlorine demand and break point chlorination with the help of a graph. Write the advantages and disadvantages of use of chlorine as a disinfectant.
  - (b) Compute the total carbonate and non-carbonate hardness of a water sample having the following analysis report;

Calcium as Ca2+ = 80 mg/L

Magnesium as Mg2+ = 36 mg/L.

Sodium as Na+=20 mg/L

Carbonate and bicarbonate as CaCO<sub>3</sub>=134 mg/L

- (a) Explain the different stability conditions in the atmosphere with respect to ideal lapse rate with the help of a diagram.
  - (b) Compute the appropriate quantity of biogas to be generated in an anaerobic digester having a flow of 3600 m<sup>3</sup>/h. assume biodegradable fraction as 0.85 and COD concentration in waste water as 5000 mg/L.

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 (a) Write the operating principle and give the labeled diagram of a condenser for control of gaseous pollutants. Also write the formula to calculate the heat exchange coefficient.

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- (b) Design an Electrostatic Precipitator to handle a quantity of 52 m³/s of air from a manufacturing industry. It contains the particles whose settling velocity is 0.13 m/s. If 99% removal efficiency is required, calculate the surface area of the ESP.
- 8. Write short notes on any two of the following:

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- (a) Explain the different stages involved in MSW landfill.
- (b) Write the properties of hazardous wastes and explain the different treatment options for hazardous wastes.
- (c) Give a flow sheet for environmental clearance process for new industries in India and discuss on it.
- (d) What is flue gas desulphurization? Discuss two popularly used compounds for reaction with SO₂ in the process.