



GANDHI INSTITUTE OF ENGINEERING AND TECHNOLOGY, ODISHA, GUNUPUR (GIET UNIVERSITY)

M.Sc. (First Semester - Regular) Examinations, February – 2025

24CHPC1001 - Organic Chemistry-I

(Chemistry)

Time: 3 hrs

Maximum: 60 Marks

Answer ALL questions

(The figures in the right hand margin indicate marks)

PART – A

(2 x 5 = 10 Marks)

Q.1. Answer **ALL** questions

- | | CO # | Blooms Level |
|---|------|--------------|
| a. Explain 2-pentene is more stable than 1-pentene. | CO1 | K2 |
| b. Phenol is acidic but cyclohexanol is not. Explain. | CO1 | K2 |
| c. Explain Inclusion Complex. | CO1 | K1 |
| d. Explain Taft Equation. | CO1 | K2 |
| e. Explain stereoselective and stereospecific reactions with example? | CO2 | K2 |

PART – B

(10 x 5 = 50 Marks)

Answer **ALL** the questions

- | | Marks | CO # | Blooms Level |
|---|-------|------|--------------|
| 2. a. Define aromatic, anti-aromatic, non-aromatic and give atleast two example for each. Explain aromaticity through benzene and non benzenoid compound. | 5 | CO1 | K2 |
| b. Discuss about the classification, structure, stability and generation of free radical. | 5 | CO1 | K1 |
| (OR) | | | |
| c. Draw the structure of cyclooctatetrene, [14]annulene and [18]annulene. Explain which one is aromatic/anti-aromatic/non-aromatic. | 5 | CO2 | K3 |
| d. Define Resonance effect, Resonance energy and Resonance Hybrid? | 5 | CO1 | K1 |
| 3.a. What are the Thermodynamic and kinetic requirements of an organic reactions? | 5 | CO2 | K2 |
| b. Derive Hammett equation to correlate substituent and reaction constant? | 5 | CO2 | K2 |
| (OR) | | | |
| c. Explain Curtin-Hammett principle. | 7 | CO2 | K3 |
| d. How do you depict the endothermic and exothermic reaction in a potential Energy Diagram ? | 3 | CO1 | K1 |
| 4.a. Discuss the optical activity of biphenyls, allenes and spiranes. | 5 | CO3 | K1 |
| b. Discuss the conformations of Cis- and trans-decaline. | 5 | CO3 | K1 |
| (OR) | | | |
| c. Write a note on Racemic Modification. | 5 | CO2 | K3 |
| d. Write short notes on E-Z Notation | 5 | CO2 | K2 |
| 5.a. Describe Shapiro reaction write with mechanism. | 5 | CO3 | K3 |
| b. Explain Mannich reaction with Mechanism. | 5 | CO3 | K3 |
| (OR) | | | |
| c. Write down Reimer-tiemann reaction with mechanism. | 5 | CO3 | K3 |
| d. Describe Claisen-Schmidt reaction with mechanism. | 5 | CO3 | K3 |
| 6.a. Give the brief idea about Semi Pinacole Pinnacolone rearrangement | 5 | CO4 | K4 |
| b. Explain Fries rearrangement with mechanism. | 5 | CO4 | K3 |
| (OR) | | | |
| c. Write a note on Sandmeyer reaction with mechanism. | 5 | CO3 | K2 |
| d. Write a note on Smiles rearrangement. | 5 | CO3 | K2 |

