

**GANDHI INSTITUTE OF ENGINEERING AND TECHNOLOGY UNIVERSITY, ODISHA, GUNUPUR
(GIET UNIVERSITY)**

M.Tech. (First Semester) Regular Examinations, February – 2025
24MTEPC11002 – Conductive and Radiative Heat Transfer
(HPTE)



Time: 3 hrs

Maximum: 60 Marks

Answer ALL questions
(The figures in the right hand margin indicate marks)

PART – A**(2 x 5 = 10 Marks)**Q.1. Answer **ALL** questions

	CO #	Blooms Level
a. Describe Fourier's law of conduction.	CO1	K1
b. Define orthotropic solids.	CO2	K1
c. Define fins (or) extended surfaces.	CO3	K1
d. Define Fin efficiency and Fin effectiveness.	CO4	K1
e. Write the differential equation governing the heat transfer in fins.	CO6	K1

PART – B**(10 x 5 = 50 Marks)**Answer **ALL** the questions

	Marks	CO #	Blooms Level
2. a. Explain in brief about the semi-infinite solid.	5	CO1	K3
b. Define isotropic and anisotropic solids? Give few differences between them.	5	CO1	K2
(OR)			
c. An aluminium rod and a copper rod of equal length 2.0 m and cross-sectional area 2 cm ² are welded together in series. One end is kept at a temperature of 10 °C and the other at 30 °C. Calculate the amount of heat taken out per second from the hot end. (Thermal conductivity of aluminium is 200 W/m °C and of copper is 390 W/m °C).	10	CO2	K3
3.a. The energy lost from a 10 cm thick slab of steel is 50 W. Assuming the temperature difference of 10.0 K, find the area of the slab. (Thermal conductivity of steel = 45 W/m K).	5	CO1	K3
b. An infinitely long pin fin, attached to an isothermal hot surface, transfers heat at a steady rate of Q ₁ , to the ambient air. If the thermal conductivity of the fin material is doubled, while keeping everything else constant, the rate of steady-state heat transfer from the fin becomes Q ₂ . Describe the ratio Q ₂ /Q ₁ ?	5	CO3	K2
(OR)			
c. A steel ball of diameter 60 mm is initially in thermal equilibrium at 1030°C in a furnace. It is suddenly removed from the furnace and cooled in ambient air at 30°C, with convective heat transfer coefficient h = 20 W/m ² K. The thermophysical properties of steel are: density ρ = 7800 kg/m ³ , conductivity k = 40 W/mK and specific heat c = 600 J/kgK. Describe the time required in seconds to cool the steel ball in the air from 1030°C to 430°C ?	10	CO4	K3
4.a. The heat loss from a fin is 6 W. The effectiveness and efficiency of the fin are 3 and 0.75, respectively. Describe the heat loss (in W) from the fin, keeping the entire fin surface at base temperature?	5	CO3	K2

- b. A 3 cm long, 2 mm x 2 mm rectangular cross-section aluminium fin [$k = 237 \text{ W/m}^\circ\text{C}$] is attached to a surface. If the fin efficiency is 65%, Describe the effectiveness of this single fin? 5 CO3 K3
- (OR)
- c. Calculate the following for an industrial furnace in the form of a black body and emitting radiation at 2500°C
Monochromatic emissive power at $1.2 \mu\text{m}$ wave length. 10 CO4 K3
- (i) Total emissive power,
(ii) The total emissive power of the furnace if it is assumed as a real surface having emissivity equal to 0.9.
- 5.a. A body takes 4 minutes to cool from 100°C to 70°C . If the room temperature is 15°C , what will be the time taken to cool from 70°C to 40°C ? 5 CO5 K3
- b. For an opaque plane surface the radiosity, irradiation and emissive power are respectively 16, 24 and 12 W/m^2 . Determine the emissivity of surface. 5 CO4 K3
- (OR)
- c. In a condenser of a power plant, the steam condenses at a temperature of 60°C . The cooling water enters at 30°C and leaves at 45°C . Describe the Logarithmic Mean Temperature Difference (LMTD) of the condenser. 5 CO3 K4
- d. Consider two infinitely long thin concentric tubes of circular cross section. If D_1 and D_2 are the diameters of the inner and outer tubes respectively, then Describe the view factor F_{22} ? 5 CO5 K3
- 6.a. Prove Kirchhoff's law of thermal radiation. 5 CO6 K2
- b. Derive relation for heat exchange between infinite parallel planes. 5 CO6 K2
- (OR)
- c. A black body at 3000 K emits radiation Calculate the following 10 CO6 K3
- (i) Wave length at which emission is maximum
(ii) Maximum emissive power
(iii) Total emissive power,

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